PATENT SPECIFICATION

(1) **1333475**

NO DRAWINGS

(22) Filed 6 May 1971 (21) Application No. 13493/71

(31) Convention Application No. P 20 24 051.1

(32) Filed 16 May 1970 in

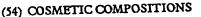
(33) Germany (DT)

(44) Complete Specification published 10 Oct. 1973

(51) International Classification A61K 7/00; C11D 1/74

(52) Index at acceptance

A5B 771 774 6B11A 6B11C 6B12B1 6B12F1 6B12F2 6B12L CCD 6B12N4 6B12NX 6B4 6C8





We, HENKEL & CIE, GMBH, a German Company, of 67, Henkelstrasse, Duesseldorf, 4000, Germany, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:-

The invention relates to a composition for 10 use in cosmetic preparations, especially cosmetic cleansing agents, containing substances for replacing the oil in the skin, based on esterification products of glycerine-ethylene oxide adducts with long-chain fatty acids.

Cosmetic cleansing agents such as shampoos, foam baths, toilet soaps and similar products cause a more or less great removal of oil from the skin on repeated use. This phenomenon is particularly pronounced when the cleansing agents are based on synthetic, surface-active substances such as alkylbenzene sulphonates, fatty alcohol sulphates, olefine sulphonates, fatty alcohol other sulphates and other surfaceactive compounds. Attempts have therefore been made to neutralise this removal of oil from the skin by replacing the oil by means of suitable additions to the cleansing agents. Allowance has in such case to be made for disadvantages, however, since the products generally concerned in replacing the oil in the skin have an unfavourable action on the foaming properties of the cleansing agents and, in cosmetic preparations based on alcoholwater mixtures, show insufficient solubility.

The present invention provides a composition for use in cosmetic preparation which comprises a surface active compound and the esterification product of an ethylene oxide addition compound produced from glycerine and 4 to 20 mol of ethylene oxide per mol of glycerine, with a fatty acid of chain length

from 8 to 18 carbon atoms in a ratio of 1 to 2 mol of fatty acid to 1 mol of glycerine-ethylene oxide addition compound as oil replacement material.

Esterification products of ethylene oxide addition compounds produced from glycerine and 7 to 15 mol of ethylene oxide per mol of glycerine with fatty acids of chain length from 8 to 18 carbon atoms in a ratio of 1 mol of fatty acid to 1 mol of glycerine-ethylene oxide addition compound are preferred as the oil replacement materials.

The preparation of the ethylene oxide addition compound as intermediate product was generally effected in known way by reacting glycerine with ethylene oxide in the desired proportions with alkaline catalysis by means of sodium ethylate. For the further treatment, the ethylene oxide addition compound obtained was reacted in the usual way with a fatty acid of chain length from 8 to 18 carbon atoms in the molar ratio of 1:1 or 1:2, using isopropyl titanate as esterification catalyst. The esterification products obtained were lightcoloured to yellowish liquids of low viscosity with an oil character to lard-like products of a faint self colour.

The quantities of oil replacement material according to the invention used in the cosmetic preparations may vary within very wide limits according to the product and its oilremoving action, and generally vary from 2 to 50% by weight, especially 5 to 25% by weight. Still higher additions are possible if the esterification products according to the invention are used at the same time in their property as surface-active substances, but in most cases this use will be of small advantage.

Oil replacement materials to be used according to the invention include, for example, esterification products from

45

			1,55	3,973
2				and of ethylene oxide with 1 mol
	the adduct of	1 mol of	glycerin	of coconut fatty acid C _{k-18} + 6 mol of ethylene oxide with 2 mol
	25	33	n	of coconut fatty acid C ₁₋₁₈ + 7 mol f ethylene oxide with 1 mol
5	33	23	ø	of coconut fatty acid C ₂₋₁₈ +7 mol of ethylene oxide with 1 mol
-	29	22	30	of tallow fatty acid + 8 mol of ethylene oxide with 1 mol
	29	23	25	of oleic acid + 9 mol of ethylene oxide with 1 mol of
10	æ	ee	39	palm kernel fatty acid + 10 mol of ethylene oxide with 1 mol
	22	22	20	of tallow fatty acid + 10 mol of ethylene oxide with 2 mol + 10 mol of ethylene oxide with 2 mol
15	23	23	22	of groundnut oil fatty acid + 12 mol of ethylene oxide with 1 mol
	23	33	*	of coconut fatty acid C ₁₋₁₈ + 15 mol of ethylene oxide with 1 mol
	23	33	23	of palm kernel fatty acid +15 mol of ethylene oxide with 2 mol
29		22	25	+ 15 mol of emylene dance

The present invention will be further described by way of illustration with reference to the following examples. Abbreviations and units used in the examples are defined as

"Acid value" is the number of mg of potassium hydroxide which are needed to neutralize the free esterified fatty acid contained in 1 g

of substance;
"Saponification value" is the number of mg
of potassium hydroxide which are needed to

completely saponify 1 g of ester;

"Hydroxyl number" is the number of mg of potassium hydroxide which are needed to neutralize the acetic acid which is absorbed by 1 g of material. The procedure used is as follows; the substance being investigated is weighed out exactly and is then acetylated by heating with crystalline acetic acid, all the OH groups present thereby being esterified. Subsequently, the acetylated portion of the material is separated by a shaking and washing procedure and the acetylated part is saponified using boiling potassium hydroxide solution. The excess potassium hydroxide which is present after the saponification steps is determined by titration with hydrochloric acid.

of palm kernel fatty acid

+ 15 mol of ethylene oxide with 2 mol
of tallow fatty acid

"EO" is the ethylene oxide group;

"WAS" is active washing substance.
All parts used throughout the examples are

by weight unless otherwise specified.

Examples

The following esterification products were used for the experiments and cosmetic preparations described below.

- (A) (1 mol of glycerine + 7.4 mol of ethylene oxide) with 1 mol of coconut fatty acid

 C. 18
 Acid value 1.0, sap. value 92, hydroxy
- value 185
 (B) (1 mol glycerine + 7.4 mol of ethylene oxide) with 1 mol of tallow fatty acid (Acid value 1.1, sap. value 83, hydroxy ratus 166
- (C) (1 mol glycerine + 10 mol of ethylene oxide) with 1 mol of tallow fatty acid Acid value 1.4, sap. value 71, hydroxy value 141

Since for cosmetic cleansing compositions the ability to combine with certain surface-active compounds is of essential importance, mixtures given in the following Table were tested.

50

70

TABLE I

Mixing component	Mixture 1	Mixture 2	Mixture 3
(A)	10	_	
(B)	_	10	
(C)			10
Sodium lauryl ether sulphate			
(2 EO) (27—28% WAS)	50	50	50
Water	40	40	40
Result	Clear homo- geneous solution	Clear homo- geneous solution	Clear homo- geneous solution

In a further experiment the foaming power of a foam bath basic recipe with additions of examined.

5

TABLE II

	Mixture	Mixture	Mixture 3	Mixture 4	Mixture 5
Components	1	2			
Sodium lauryl ether sulphate (2 EO) (27—28% WAS)	60	60	60	60	60
Sodium lauryl sulphate (over 90% WAS)	5	5	5	5	5
Isopropyl		5		-	
myristate	<u>-</u>		5	_	-
(A)	_	_		5	-
(B)	_	_		_	5
(C)	_		20	30	30
Water	35	30	30	clear	clear
Appearance	clear	turbid deposit formed	dear		
Foaming power					
Initial volume in	ml 1 minute	after beating	200	340	240
0.5 g/litre	260	170	300	490	400
1.0 g/litre	510	250	490	-	540
2 0 g/litre	660	330	610	640	
2.0 g/ == 1	of the volume	of foam in ml/m	inute		
	2.0	2.5	4.0	4.5	1.5
0.5 g/litre		3.5	7.0	8.0	4.0
1.0 g/litre	5.5	6.5	9.5	3.0	6.5
2.0 g/litre	8.0				

The foaming power of the individual mixtures was measured in the foam-beating machine according to DIN (Deutsche Industrie Norm) classification No. 53,902, in which the volume of foam was measured. The figures for the foam were taken at 45°C in water of 10° German hardness after 30 beats. The measurement was taken 1 minute after the 10 end of the beating and 21 minutes after the end of the beating. The breakdown of the volume of foam was calculated in ml/minute from the decrease of the volume of foam in

20 minutes. The amounts given in g/litre relate to the respective mixture.

As may be seen from the above Table, the foam values of the mixtures containing oil replacement means according to the invention are substantially better than when isopropyl myristate is used as oil replacement materials, and are scarcely inferior to the figures for a mixture of pure detergent substances.

A few formulations for cosmeric preparations containing oil replacement materials according to the invention are given below.

Clear Shampoo

Sodium lauryl sulphate (2EO)	40	parts	bу	weight
(27—28% WAS)	6	25	23	23
(27—28% WAS) Coconut fatty acid diethanolamide	10	23	20	20
Oil replacement means (A)	44	20	20	20
Water	• • •	~		

Shampoo for dry hair

Sodium lauryl ether sulphate	20	parts	by we	ight
(2 EO) (27—28% WAS) Sodium lauryl sulphate (90% WAS)	5	20	23	39
Coconut fatty acid menoethanol-	3 5	33 33	29 29	13 23
amide paste 30%	0.5 25.0	20	נג נג	33
Oil replacement means (B) Water	41.5))))	ນ	23

Foam bath

Sodium lauryl ether sulphate (2 EO) (27—28% WAS) Sodium lauryl sulphate	30 1	arts	by '	weight
Sodium lauryl sulphate	15		33	20
(90% WAS) Coconut fatty acid diethanolamide	5	99	33	20
Coconut latty and dictions	5	33	33	30
Pine-needle oil	10	23	23	33
Oil replacement means (C)	35	93	23	23
Water		-		

Hair wash

TIME WALL		_	
_	60.0 pa	rts by	weignt
Isopropanol	0.2		33
Menthol	0.05	, 25	29
Calcium pantothenate	0.30	, 33	33
Vitamin H	0.10	دد و	2)
Inositol	0.50	. א	30
Perfume	5.00	23 25	29
Oil replacement means (C)	33.85	29 33	33
Water	_	-	

After-shave lotion

ight
ט
20
20
23
- 10
ນ
2 0
23
20 20
~
1

Sun tan cream

Colloidally dispersed mixture of 90 parts of cetylstearyl alcohol and 10 parts of sodium lauryl 10.0 parts by weight sulphate 10 2-Octyldodecanol 5 33 Groundnut oil Light-protective means 20 27 Oil replacement means (B) 53 Water

The oil replacement compositions according to the invention can be used particularly cleansing cosmetic in advantageously means, because they do not exert any appreciable influence on the foaming power of the surface-active products, and because they already have a good solubility in alcoholwater mixtures.

WHAT WE CLAIM IS:-10

1. A composition, for use in cosmetic preparations, which comprises a surface active compound and the esterification product of an ethylene oxide addition compound produced 15 from glycerine and 4 to 20 mol of ethylene oxide per mol of glycerine with a fatty acid of chain length from 8 to 18 carbon atoms in a ratio of 1 to 2 mol of fatty acid to 1 mol of glycerine-ethylene oxide addition compound as oil replacement material.

2. A composition according to claim 1, in which the ratio of glycerine to ethylene oxide in the addition compound is from 7 to 15 mols

of ethylene oxide per mol of glycerine.

3. A composition according to claim 1 or 2 wherein the fatty acids of chain length from 8 to 18 carbon atoms are in a ratio of 1 mol of fatty acid to 1 mol of the glycerine-ethylene oxide addition compound.

4 A composition according to claims 1 to 3, containing from 2 to 50% by weight of the oil replacement material.

5. A composition according to claim 1 to 4 containing from 5 to 25% by weight of the oil replacement material.

6. A composition according to claim 1 substantially as hereinbefore described with reference to and as illustrated in the foregoing examples.

7. A cosmetic preparation whenever containing a composition as claimed in any one

of claims 1 to 6.

W. P. THOMPSON & CO.,

12, Church Street, Liverpool, L1 3AB. Chartered Patent Agents.

Printed for Her Majesty's Stationery Office by the Courier Press, Learnington Spa, 1973. Published by the Patent Office, 25 Southampton Buildings, London, WC2A 1AY, from which copies may be obtained.